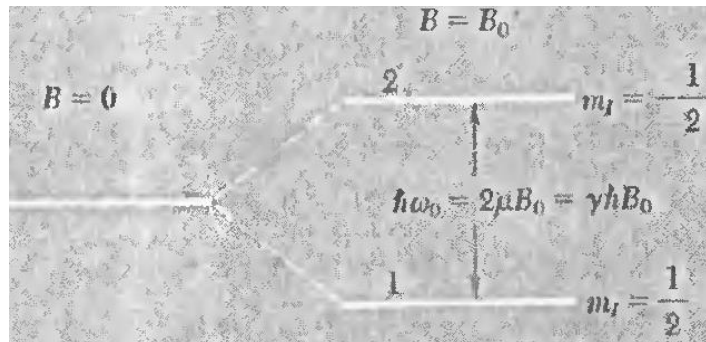
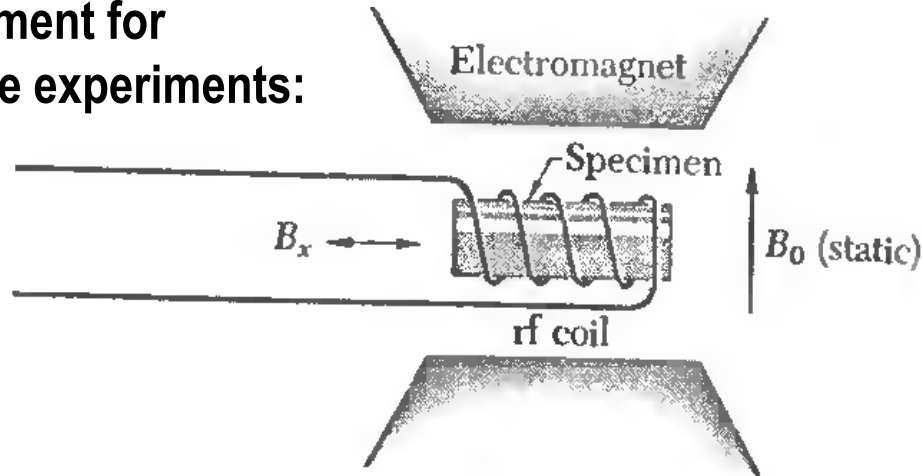


# Magnetic resonance (2)

Schematic arrangement for magnetic resonance experiments:

To rf supply and circuit for measuring inductance and losses.



Energy level splitting of a nucleus of spin 1/2 in a static magnetic field  $B_0$ .

Magnetic moment of a nucleus  $\mu = \gamma\hbar I$

The energy of interaction with the applied magnetic field is  $U = -\mu \cdot B_a$

For  $I=1/2$  the magnetic resonance frequency

$\omega_0 = \gamma B_0$  where the constant  $\gamma$  depends on the magnetic moment

For proton

$$\gamma = 2.675 \times 10^4 \text{ s}^{-1} \text{ gauss}^{-1} = 2.675 \times 10^8 \text{ s}^{-1} \text{ tesla}^{-1}$$

Table 1 Nuclear magnetic resonance data

For every element the most abundant magnetic isotope is shown. After Varian Associates NMR Table.

Table 1 Nuclear magnetic resonance data																			He <sup>3</sup>			
H <sup>1</sup>																	1/2					
99.98																	10 <sup>-6</sup>					
2.792																	-2.127					
For every element the most abundant magnetic isotope is shown. After Varian Associates NMR Table.																						
Li <sup>7</sup>		Be <sup>9</sup>															B <sup>11</sup>	C <sup>13</sup>	N <sup>14</sup>	O <sup>17</sup>	F <sup>19</sup>	Ne <sup>21</sup>
3/2		3/2															3/2	1/2	1	5/2	1/2	3/2
92.57		100.															81.17	1.108	99.64	0.04	100.	0.257
3.256		-1.177															2.688	0.702	0.404	-1.893	2.627	-0.662
Na <sup>23</sup>		Mg <sup>25</sup>		Most abundant isotope with nonzero nuclear spin													Al <sup>27</sup>	Si <sup>29</sup>	P <sup>31</sup>	S <sup>33</sup>	Cl <sup>35</sup>	Ar
3/2		5/2		Nuclear spin; in units of $\hbar$													5/2	1/2	1/2	3/2	3/2	
100.		10.05		Natural abundance of isotope, in percent													100.	4.70	100.	0.74	75.4	
2.216		0.855		Nuclear magnetic moment, in units of $e\hbar/2M_p c$													3.639	0.555	1.131	0.643	0.821	
K <sup>39</sup>	Ca <sup>43</sup>	Sc <sup>45</sup>	Ti <sup>47</sup>	V <sup>51</sup>	Cr <sup>53</sup>	Mn <sup>55</sup>	Fe <sup>57</sup>	Co <sup>59</sup>	Ni <sup>61</sup>	Cu <sup>63</sup>	Zn <sup>67</sup>	Ga <sup>69</sup>	Ge <sup>73</sup>	As <sup>75</sup>	Se <sup>77</sup>	Br <sup>79</sup>	Kr <sup>83</sup>					
3/2	7/2	7/2	5/2	7/2	3/2	5/2	1/2	7/2	3/2	3/2	5/2	3/2	9/2	3/2	1/2	3/2	9/2					
93.08	0.13	100.	7.75	~100.	9.54	100.	2.245	100.	1.25	69.09	4.12	60.2	7.61	100.	7.50	50.57	11.55					
0.391	-1.315	4.749	0.787	5.139	0.474	3.461	0.090	4.639	0.746	2.221	0.874	2.011	0.877	1.435	0.533	2.099	-0.967					
Rb <sup>85</sup>	Sr <sup>87</sup>	Y <sup>89</sup>	Zr <sup>91</sup>	Nb <sup>93</sup>	Mo <sup>95</sup>	Tc	Ru <sup>101</sup>	Rh <sup>103</sup>	Pd <sup>105</sup>	Ag <sup>107</sup>	Cd <sup>111</sup>	In <sup>115</sup>	Sn <sup>119</sup>	Sb <sup>121</sup>	Te <sup>125</sup>	I <sup>127</sup>	Xe <sup>129</sup>					
5/2	9/2	1/2	5/2	9/2	5/2		5/2	1/2	5/2	1/2	1/2	9/2	1/2	5/2	1/2	5/2	1/2					
72.8	7.02	100.	11.23	100.	15.78		16.98	100.	22.23	51.35	12.86	95.84	8.68	57.25	7.03	100.	26.24					
1.348	1.089	0.137	1.298	6.144	0.910		-0.69	0.088	-0.57	-0.113	-0.592	5.507	-1.041	3.342	-0.882	2.794	-0.773					
Cs <sup>133</sup>	Ba <sup>137</sup>	La <sup>139</sup>	Hf <sup>177</sup>	Ta <sup>181</sup>	W <sup>183</sup>	Re <sup>187</sup>	Os <sup>189</sup>	Ir <sup>193</sup>	Pt <sup>195</sup>	Au <sup>197</sup>	Hg <sup>199</sup>	Tl <sup>205</sup>	Pb <sup>207</sup>	Bi <sup>209</sup>	Po	At	Rn					
7/2	3/2	7/2	7/2	7/2	1/2	5/2	3/2	3/2	1/2	3/2	1/2	1/2	1/2	9/2								
100.	11.32	99.9	18.39	100.	14.28	62.93	16.1	61.5	33.7	100.	16.86	70.48	21.11	100.								
2.564	0.931	2.761	0.61	2.340	0.115	3.176	0.651	0.17	0.600	0.144	0.498	1.612	0.584	4.039								
Fr	Ra	Ac																				
			Ce <sup>141*</sup>	Pr <sup>141</sup>	Nd <sup>143</sup>	Pm	Sm <sup>147</sup>	Eu <sup>153</sup>	Gd <sup>157</sup>	Tb <sup>159</sup>	Dy <sup>163</sup>	Ho <sup>165</sup>	Er <sup>167</sup>	Tm <sup>169</sup>	Yb <sup>173</sup>	Lu <sup>175</sup>						
			7/2	5/2	7/2		7/2	5/2	3/2	3/2	5/2	7/2	7/2	1/2	5/2	7/2						
			—	100.	12.20		15.07	52.23	15.64	100.	24.97	100.	22.82	100.	16.08	97.40						
			0.16	3.92	-1.25		-0.68	1.521	-0.34	1.52	-0.53	3.31	0.48	-0.20	-0.677	2.9						
			Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr						

# Equations of motion for magnetic moment

The time evolution of a spin vector  $\mathbf{I}$  is given by  $\hbar d\mathbf{I}/dt = \boldsymbol{\mu} \times \mathbf{B}_a$

or for its magnetic moment  $d\boldsymbol{\mu}/dt = \gamma\boldsymbol{\mu} \times \mathbf{B}_a$

The evolution of a total magnetization in external field  $d\mathbf{M}/dt = \gamma\mathbf{M} \times \mathbf{B}_a$

**In thermal equilibrium the population of two energy levels is given by the ratio:**

$$(N_2/N_1)_0 = \exp(-2\mu B_0/k_B T)$$

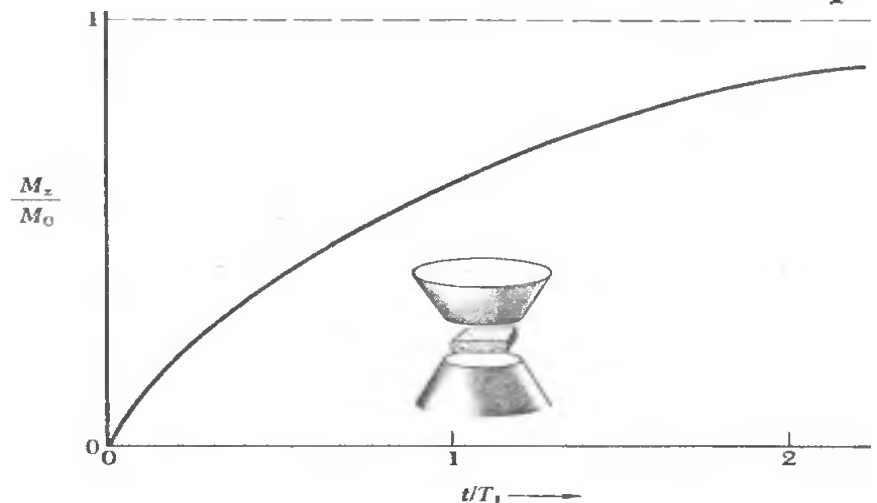
**and the total magnetization is given by**  $M_0 = N\mu \tanh(\mu B/k_B T)$

When magnetization is not in equilibrium, it approaches equilibrium  $\frac{dM_z}{dt} = \frac{M_0 - M_z}{T_1}$  at a rate proportional to the departure from the equilibrium value  $M_0$ :

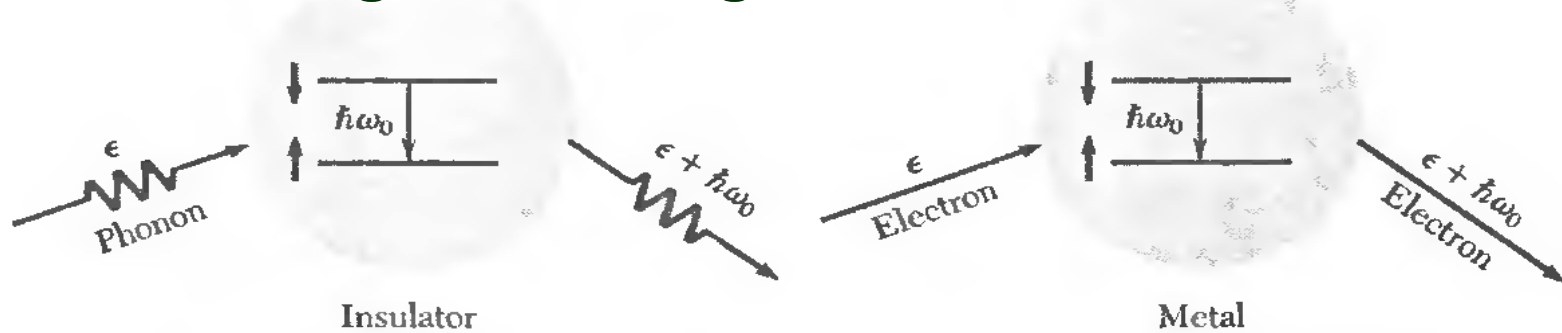
This linear differential equation can be solved easily and gives

$$M_z(t) = M_0[1 - \exp(-t/T_1)]$$

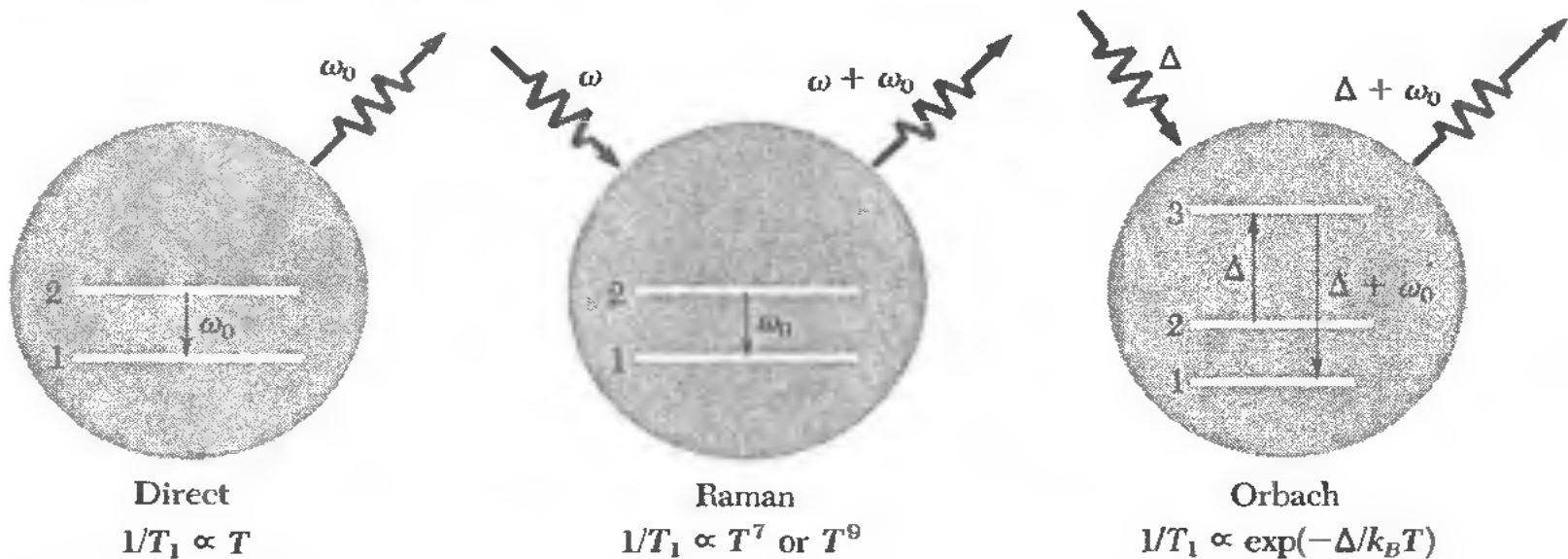
$T_1$  is called the longitudinal relaxation time or the spin-lattice relaxation time.



# Longitudinal magnetization relaxation



**Figure 4a** Some important processes that contribute to longitudinal magnetization relaxation in an insulator and in a metal. For the insulator we show a phonon scattered inelastically by the spin system. The spin system moves to a lower energy state, and the emitted phonon has higher energy by  $\hbar\omega_0$  than the absorbed phonon. For the metal we show a similar inelastic scattering process in which a conduction electron is scattered.



**Figure 4b** Spin relaxation from  $2 \rightarrow 1$  by phonon emission, phonon scattering, and a two-stage phonon process. The temperature dependence of the longitudinal relaxation time  $T_1$  is shown for the several processes.

# Longitudinal magnetization relaxation (experiment)

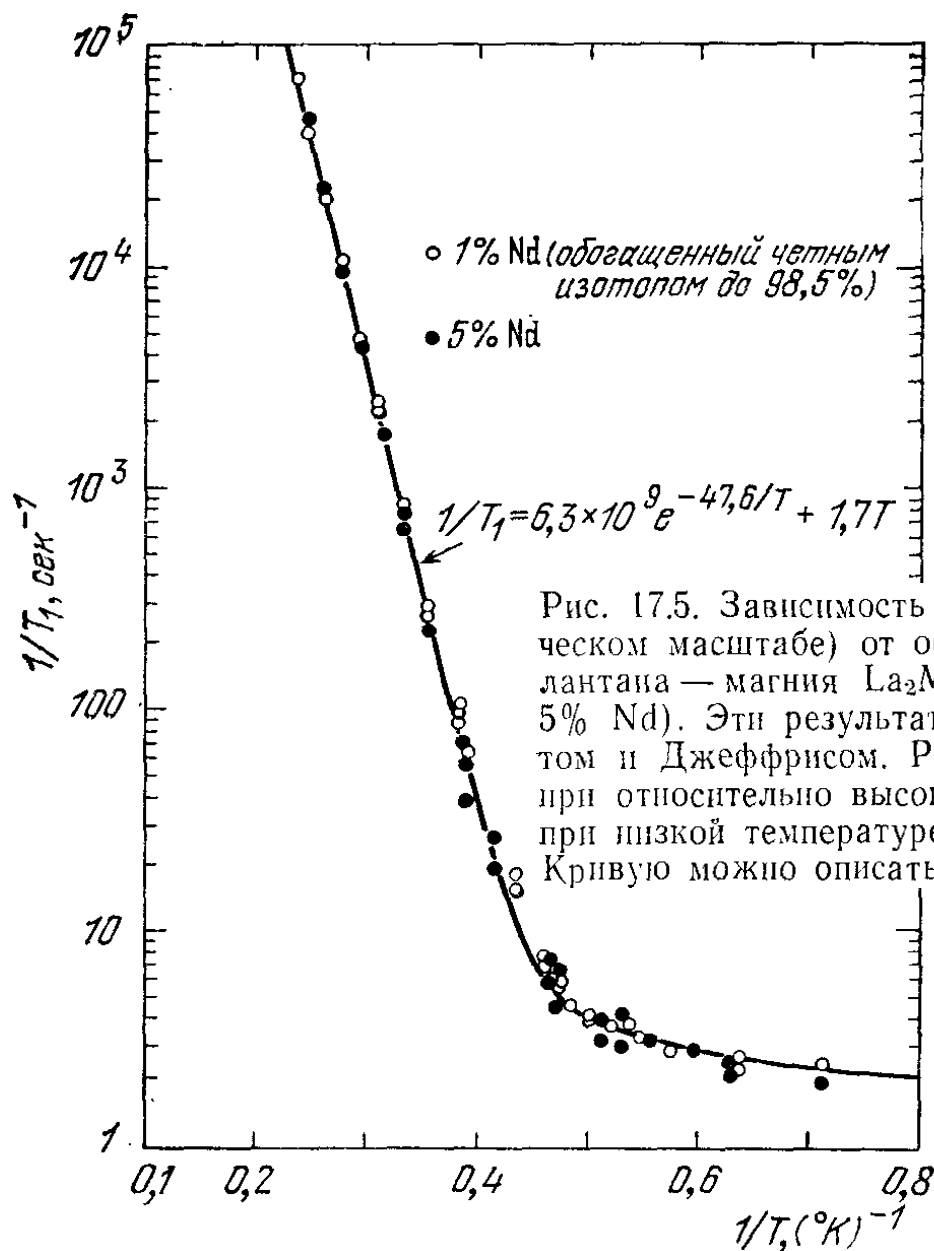


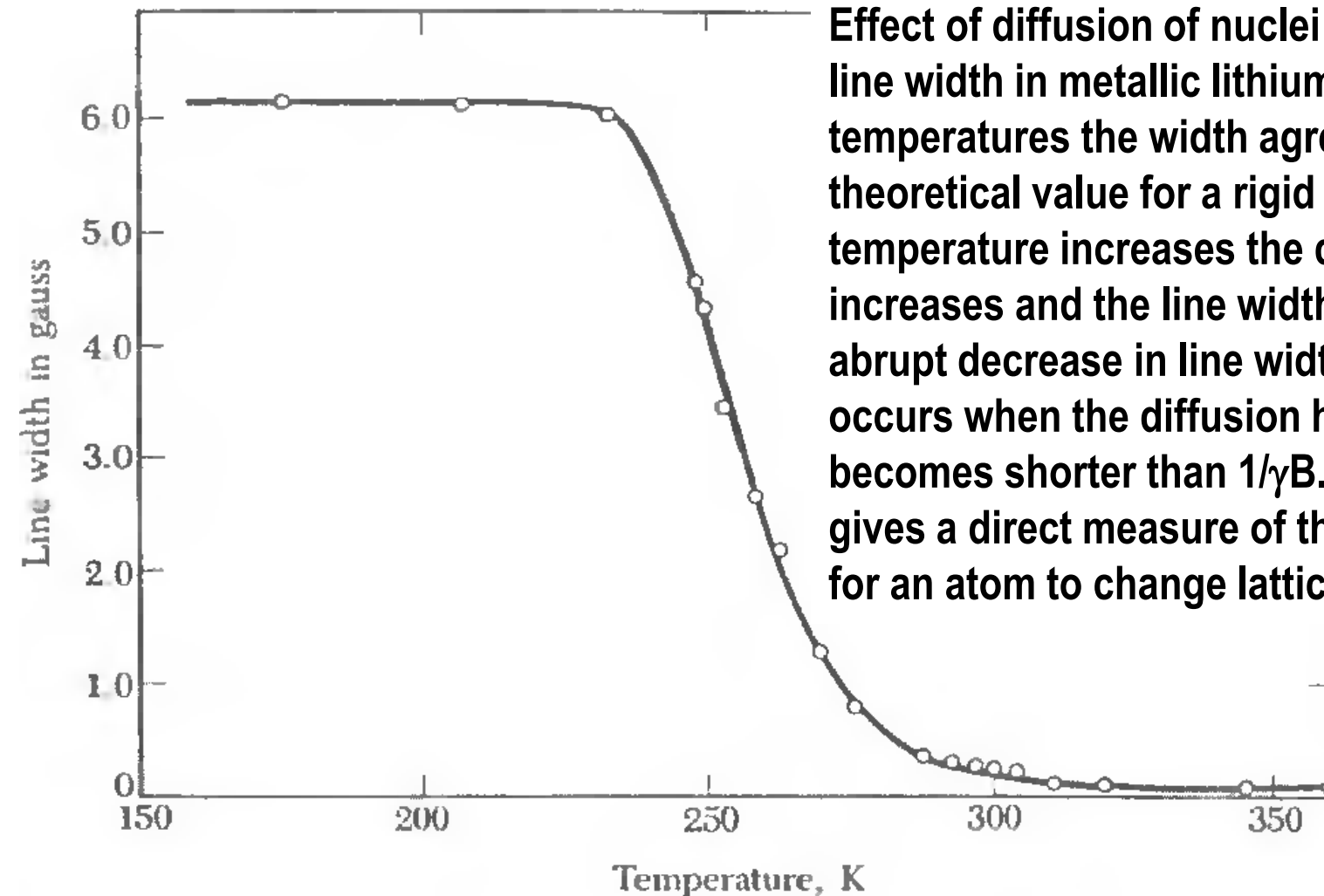
Рис. 17.5. Зависимость быстроты продольной релаксации  $1/T_1$  (в логарифмическом масштабе) от обратной температуры для кристалла двойного нитрата лантана — магния  $\text{La}_2\text{Mg}_3(\text{NO}_3)_{12} \cdot 24\text{H}_2\text{O}$  с примесью неодима (1% Nd и 5% Nd). Эти результаты получены в интервале температур 1,4—4,3°K Скоттом и Джеффрисом. Результаты убедительно указывают на процесс Орбаха при относительно высоких температурах и на прямой однофононный процесс при низкой температуре. Частота  $\nu = 9,37$  ГГц, магнитное поле  $B = 2,48$  кГс. Кривую можно описать аналитически:  $1/T_1 = 6,3 \cdot 10^9 e^{-47,6/T} + 1,7T$ .

# Motional Narrowing of NMR line

The effective magnetic field due to magnetic dipole-dipole interaction is

$$\Delta B = \frac{3(\mu_2 \cdot r_{12})r_{12} - \mu_2 r_{12}^2}{r_{12}^5}$$

It effectively averages to almost zero when the atoms move fast enough.



Effect of diffusion of nuclei on the  $\text{Li}^7$  NMR line width in metallic lithium. At low temperatures the width agrees with the theoretical value for a rigid lattice. As the temperature increases the diffusion rate increases and the line width decreases. The abrupt decrease in line width above  $T = 230$  K occurs when the diffusion hopping time becomes shorter than  $1/\gamma B$ . The experiment gives a direct measure of the hopping time for an atom to change lattice sites.



# The Bloch equations for magnetic moment evolution with time

$$dM_x/dt = \gamma(\mathbf{M} \times \mathbf{B})_x - M_x/T_2 ;$$

$$dM_y/dt = \gamma(\mathbf{M} \times \mathbf{B})_y - M_y/T_2 ;$$

$$dM_z/dt = \gamma(\mathbf{M} \times \mathbf{B})_z + (M_0 - M_z)/T_1$$

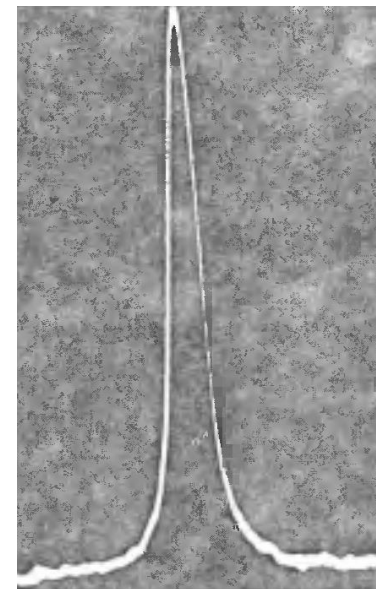
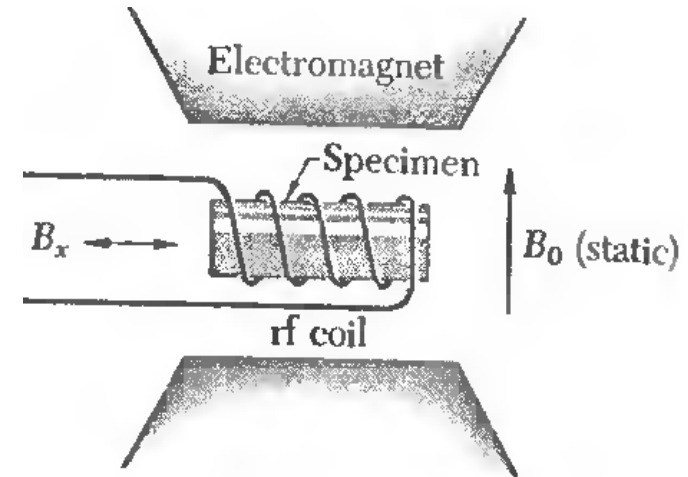
$T_2$  is called the **transverse relaxation time**

The transverse spin relaxation does not need energy change, => often  $T_2 \ll T_1$

Solving the Bloch equations we find the  
resonance power absorption in NMR  
experiments with resonance frequency  $\omega_0$   
determined by Zeeman energy splitting in  $B_0$ :

$$\mathcal{P}(\omega) = \frac{\omega \gamma M_z T_2}{1 + (\omega_0 - \omega)^2 T_2^2} B_1^2$$

The resonance half-width  $(\Delta\omega)_{1/2} = 1/T_2$



**Proton  
resonance  
absorption  
in water.**